

Chapter 9 Mixed Review Stoichiometry

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CHAPTER 9 REVIEW Stoichiometry CHAPTER 9 REVIEW. Stoichiometry. MIXED REVIEW. SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: C3H4(g) + x. O2(g) (3CO2(g) + 2H2O(g) a. What is the value of the coefficient . x. in this equation? b. What is the molar mass of C3H4? c. How many moles are in an 8.0 g sample of C3H4?

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Stoichiometry b. Theoretically, how many moles of NH3 will be produced? PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N2 are mixed with 12.0 mol of H2 according to the ...

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Reaction stoichiometry, the subject of this chapter, is based on chemical equations and the law of conservation of mass. All reaction stoichiometry calculations start with a balanced chemical equation. This equation gives the ... 290 Chapter 9 DO NOT EDIT--Changes must be made through ¶File info¶ ...

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