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Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q_1 and Q_2 separated by a distance r is given by $PE = \frac{1}{4\pi\epsilon_0} \frac{Q_1 Q_2}{r}$ First we consider the interaction between Na^+ ion and $6Cl^-$ ions at distance r Applying Coulomb's law we have $Q_1 Q_2 = (-6e)(+e) = -6e^2$ $PE_1 = \frac{1}{4\pi\epsilon_0} \frac{-6e^2}{r}$ Similarly, we now consider $12 Na^+$ ions as second nearest neighbors at a distance r $Q_1 Q_2 = (+12e)(+e) = 12e^2$ $PE_2 = \frac{1}{4\pi\epsilon_0} \frac{12e^2}{r}$ and Cl^- ...

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